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triphenylphosphineacetyl palladium; copper complex salts of diphenanthroline, substituted diphenanthroline, 2,2' : 6',2''-terpyridine, pyridinimine, or cross-linked aliphatic diamine, and copper complexes such as acetyl[4-4'-di(5-nonyl)-2,2'-bipyridine] copper, hexafluorophosphine-di[4-4'-di(5-nonyl)-2,2'-bipyridine] copper, copper thiocyanate, bipyridine copper with coordinated O, S, Se, or Te, and copper with coordinated alkylbipyridinylamine, alkyl substituted tripyridine, di(alkylamino)alkylpyridine, iminodipyridine, ethylenediaminedipyridine, or tri(pyridinylmethyl)amine. Of these, dicarbonylcyclopentadienyl iron (I) (II) iodide, dicarbonylcyclopentadienyl ruthenium(II) iodide, and carbonylcyclopentadienyl nickel(II) iodide and the like are preferred. These transition metal complexes can be used either singularly, or in combinations of two or more different complexes.--

Please replace the paragraph starting from page 41, line 5 with the following amended paragraph:

--Specific examples of suitable electrolyte salts include conventional alkali metal salts such as LiCF_3SO_3 , $\text{LiN}(\text{CF}_3\text{SO}_2)_2$, $\text{LiC}(\text{CF}_3\text{SO}_2)_3$, $\text{LiC}(\text{CH}_3)(\text{CF}_3\text{SO}_2)_2$, $\text{LiCH}(\text{CF}_3\text{SO}_2)_2$, $\text{LiCH}_2(\text{CF}_3\text{SO}_2)_2$, $\text{LiC}_2\text{F}_5\text{SO}_3$, $\text{LiN}(\text{C}_2\text{F}_5\text{SO}_2)_2$, $\text{LiN}(\text{CF}_3\text{SO}_2)_2$, $\text{LiB}(\text{CF}_3\text{SO}_2)_2$, LiPF_6 , LiSbF_6 , LiClO_4 , LiI , LiBF_4 , LiSCN , LiAsF_6 , NaCF_3SO_3 , NaPF_6 , NaClO_4 , NaI , NaBF_4 , NaAsF_6 , KCF_3SO_3 , KPF_6 , KI , LiCF_3CO_3 , NaClO_3 , NaSCN , KBF_4 , KPF_6 , $\text{Mg}(\text{ClO}_4)_2$, and $\text{Mg}(\text{BF}_4)_2$, and these electrolyte salts may be used either singularly, or in mixtures of two or more different salts. Of the above electrolyte salts, lithium salts are particularly preferred.--

Please replace the paragraph starting from page ~~48~~⁵⁰, line ~~1~~¹⁶ with the following amended paragraph:

--To 47 g of toluene under an argon atmosphere were added 0.02 g (0.02 mmol) of chloropentamethylcyclopentadienyl-bis(triphenylphosphine)ruthenium, 17.8 g (0.1 mmol) of the poly-(styrene-b-AME-400) polymer obtained in (2) above, 3.7 g (18 mmol) of isobornyl acrylate, and 0.57 g (5 mmol) of n-octane, and following mixing to generate a homogeneous solution, 0.03 g (0.2 mmol) of di-n-butylamine was added, and the resulting mixture was heated to 80°C under constant stirring to initiate the copolymerization reaction. Following reaction for 20 hours from the

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6/2/09